

# Chemistry 2nd Semester Final Exam Study Guide

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Convert grams to moles to atoms. Divide by the molar mass, then multiply by  $6.022 \times 10^{23}$

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The following are topics you should know about from each chapter covered so far second semester the final exam Naming Acids and Chemistry Final Exam Review

May 23, 2014 Questions located at: This video discusses all of the topics that one would expect to find on the second semester

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Chemistry 2nd Semester Exam Study Guide. The SI base unit for the amount of substance is the mole. The SI base unit for the mass of a substance gram.

Chemistry Final. Mrs. O'Gorman. Semester 1. 2010. 1. The state of matter for an object that has definite shape and definite volume is

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Second Semester Topics. Review all of the questions in the review packet for the following topics: The Mole (omit 36, 37) Chemistry Final Exam Study Guide.

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